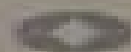


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web reduce in volume as a result of applied pressure contraction reduce in volume as a result of cooling compounds a substance defined by a particle composed of two or more different kinds of atoms elements a fundamental substance that cannot be broken into a simpler substance by chemical or physical processes

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web mar 6 2017 1 the chemical reaction $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ is a synthesis reaction decomposition reaction single displacement reaction double displacement reaction combustion reaction you can recognize a decomposition or analysis reaction by noticing a compound or molecule breaks into smaller chemical species 2 the chemical reaction

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web for each reaction in part i and part ii record your observations molecular equation total ionic equation and net ionic equation make sure to write any evidence of any evidence of a chemical reaction with sufficient detail to help you

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web post test chemical reactions 5 0 1 review identify the type of reaction represented by each equation a $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$ b $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ equation a represents a reaction and equation b represents a

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web to perform and observe the results of a variety of chemical reactions to become familiar with the observable signs of chemical reactions to identify the products formed in chemical reactions and predict when a reaction will occur to write balanced equations for the reactions studied

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web post test questions when two or more atoms or reactants bond to form a new more complex structure the reaction is called a if a patient's blood pH moves from 7 42 to 7 7 which of the following is true the patient has a condition called alkalosis the patient has a condition called acidosis a small change in pH reflects a small change

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web what type of chemical reaction is shown $\text{AB} + \text{CD} \rightarrow \text{AD} + \text{BC}$ decomposition the reaction $\text{Pb} + \text{NO}_3 \rightarrow 2\text{KI} + \text{PbI}_2 + 2\text{KNO}_3$ is a double

replacement reaction the reaction $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ is a synthesis

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 $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
 which chemical symbols will complete the equation for this single displacement reaction
 $2\text{NaBr} + \text{Cl}_2 \rightarrow$ balance $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ and more