

Pencil Electrolysis

Water is often considered to be a rather stable compound; that is, adding energy in the form of heat only causes a change in its state, but the steam that results is still H_2O . Nevertheless, you can use household materials to decompose water into its elements!

Materials

- 9-volt battery
- 2 wooden pencils
- electrical tape
- 2 4- to 6-inch pieces of small gauge insulated wire (two different colors are preferable)
- masking tape
- marker
- shallow cup or bowl
- table salt or Epsom salt
- water
- 1 of the following combinations (to make an acid-base indicator):
 - * red cabbage and either a blender or a stove and pan.
 - * Ex-Lax® tablet and spoon

Getting Ready

Step 1 Electrolysis apparatus: Sharpen both ends of the pencils. Use electrical or masking tape to attach the battery between the pencils. (See Figure 1.) Strip two centimeters of the insulation from both ends of each of the two pieces of wire. Wrap one end of each piece of wire to a terminal of the battery (Red traditionally goes with positive, and black or green goes with negative). Make sure that the stripped ends are in contact with the terminals, then secure with electrical tape. Wrap the other end of each wire to the sharpened tip of a pencil. Make sure that the stripped ends are in contact with the lead (graphite) of the pencil tips, then secure with electrical tape. With masking tape and a marker, label which wire goes to the positive terminal and which goes to the negative terminal.

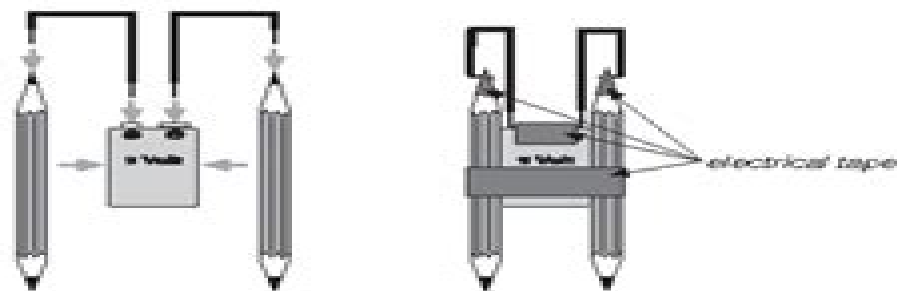


Figure 1: Assemble the pencil electrolysis apparatus.

Step 2 Electrolyte solution: Pour about 1 cm of water into the cup or bowl. Stir in table salt or Epsom salt until no more will readily dissolve.

Electrochemistry Pencil Electrolysis Lab Answers

Roman Wölfel



Electrochemistry Pencil Electrolysis Lab Answers:

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