

App 4: Thermodynamics – 12 August 2018, 12:00pm-12:30pm (with special discussion Tuesday 13:00-13:15, after lecture)

Remembering important facts about...

- Remembering systems in equilibrium
- Check the conditions: isolated, is., is. for a thermodynamic system
- System has 2 given values: temperature and pressure
- Change a thermodynamic system - remember that thermodynamic systems are equilibrium systems (pressure and temperature homogeneous, pressure uniform, pressure uniform)
- Change a thermodynamic system for a thermodynamic system
- Remember the thermodynamic system is is.
- Remember that thermodynamic systems are is.

Example: Thermodynamic System with 2 parameters (Temperature, Pressure, Volume)

Consider the following system: a gas in a cylinder with a piston. The system is in equilibrium with the surroundings. The system is in equilibrium with the surroundings.

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1.  $U = U(T, P, V)$ 
2.  $U = U(T, P, V)$ 
3.  $U = U(T, P, V)$ 
4.  $U = U(T, P, V)$ 
5.  $U = U(T, P, V)$ 
6.  $U = U(T, P, V)$ 

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Consider the following system: a gas in a cylinder with a piston.

When the gas is in equilibrium with the surroundings, the system is in equilibrium with the surroundings. When the gas is in equilibrium with the surroundings, the system is in equilibrium with the surroundings. When the gas is in equilibrium with the surroundings, the system is in equilibrium with the surroundings.

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5.  $U = U(T, P, V)$ 
6.  $U = U(T, P, V)$ 

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10. The following equation is a $U = U(T, P, V)$ equation for the thermodynamic system $U = U(T, P, V)$.

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11. The following equation is a $U = U(T, P, V)$ equation for the thermodynamic system $U = U(T, P, V)$.

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Pamela Mills,Amina El-Ashmawy



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