

CASCADED MC QUESTIONS: GENERAL CHEMISTRY II

PUT THE FOLLOWING QUESTIONS FROM SCHOOL

1. Which among the intermolecular forces is the weakest?
 - A. Hydrogen Bond
 - B. Ion-Dipole Forces
 - C. Dipole-Dipole Forces
 - D. Dispersion Forces
2. In which of the following would dipole-dipole forces exist?
 - A. Butane
 - B. Methane
 - C. CH_2Cl_2
 - D. CH_3OH
3. The vapor pressure of a liquid at 100 °C is 1.00 atm. What would be the vapor pressure at 50 °C?
 - A. 1.00 atm
 - B. Greater than 1.00 atm
 - C. Less than 1.00 atm
 - D. Cannot be determined
4. Which of the following statements about the following process will be correct at equilibrium?

$\text{H}_2\text{O}(l) \rightleftharpoons \text{H}_2\text{O}(g)$

 - A. All the water will evaporate
 - B. All water will condense
 - C. It should take some time to reach equilibrium at constant
 - D. It depends on the nature of water
5. Which of the following would have the lowest boiling point?
 - A. H_2O
 - B. CH_3OH
 - C. H_2S
 - D. CH_3SH
6. Which of the following solvents would be the best solvent for NaCl ?
 - A. Benzene
 - B. water
 - C. CH_2Cl_2
 - D. ether
7. Which of the following describes Henry's Law?
 - A. Partial pressure increases with the solubility of gases.
 - B. Partial pressure increases with the solubility of gases.
 - C. Partial pressure increases with the solubility of solids.
 - D. Partial pressure increases with the solubility of solids.
8. How much of a substance is dissolved in 1.00 L of solution if the substance has a molar mass of 100 g/mol and the solution has a concentration of 1.00 M?
 - A. 1.00 mol
 - B. 1.00 g
 - C. 1.00 mol/L
 - D. enough water to make 1.00 L of solution
9. Water's vapor pressure at 20 °C is 17.5 mm Hg. The temperature is held constant while adding NaCl resulting in a solution of 0.100 M. Determine the vapor pressure of the solution.
 - A. 17.5 mm Hg
 - B. 1.75 mm Hg
 - C. 1.75 mm Hg
 - D. 1.75 mm Hg
10. Which of the following is not a strong electrolyte?

NaCl , H_2SO_4 , CH_3COOH , H_2O

 - A. NaCl
 - B. H_2SO_4
 - C. CH_3COOH
 - D. H_2O

General Chemistry Diagnostic Test

J Elliott



General Chemistry Diagnostic Test:

General Chemistry Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996 General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 *Resources in Education*, 1997 *Teaching Innovation in University Education: Case Studies and Main Practices* Saura, Jose Ramon, 2022-06-17 In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based methods Teaching Innovation in University Education Case Studies and Main Practices features the major practices and case studies of teaching innovation developed in recent years at universities It is a source on study cases focused on teaching innovation methodologies as well as on the identification of new technologies that will help the development of initiatives and practices focused on teaching innovation at higher education institutions Covering topics such as didactic strategics service learning and technology based gamification this premier reference source is an indispensable resource for pre service teachers lecturers students faculty administrators libraries entrepreneurs researchers and academicians **International Record of Medicine and General Practice Clinics** Frank Pierce Foster, 1920 **Science Inquiry, Argument and Language**, 2019-02-18 Science Inquiry Argument and Language describes research that has focused on addressing the issue of embedding language practices within science inquiry through the use of the Science Writing Heuristic approach In recent years much attention has been given to two areas of science education scientific argumentation and science literacy The research into scientific argument have adopted different orientations with some focusing on science argument as separate to normal teaching practices that is teaching students about science argument prior to using it in the classroom context while others have focused on embedding science argument as a critical component of the inquiry process The current emphasis on science literacy has emerged because of greater understanding of the role of language in doing and reporting on science Science is not viewed as being separate from language and thus there is emerging research emphasis on how best to improving science teaching and learning through a language perspective Again the research orientations are parallel to the research on scientific argumentation in that the focus is generally between instruction separate to practice as opposed to embedding language practices within the science classroom context **Tests and Measurements** Henry Lester Smith, Wendell William Wright, 1928 **Chemists' Guide to Effective Teaching** Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 Part of the Prentice Hall Series in Educational Innovation for Chemistry this unique book is a collection of information examples and references on learning theory teaching methods and pedagogical issues related to teaching chemistry to

college students In the last several years there has been considerable activity and research in chemical education and the materials in this book integrate the latest developments in chemistry Each chapter is written by a chemist who has some expertise in the specific technique discussed has done some research on the technique and has applied the technique in a chemistry course

Journal of Chemical Education ,1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Tests and Measurements in High School Instruction Giles Murrell Ruch,George Dinsmore Stoddard,1927 **Program and Abstracts** American Chemical Society. Northwest Regional Meeting,1989 *Survival Handbook for the New Chemistry Instructor* Diane M. Bunce,Cinzia M. Muzzi,2004 This book provides an overview of the issues facing new chemistry faculty in preparation for teaching Serving as a reference to answer specific questions new chemistry faculty encounter this book is comparable to sitting down with a colleague in the department and talking through some ideas or gaining some pointers on how to avoid common pitfalls It is the one single place new chemistry faculty can go to find practical information on how to teach and how to prepare for teaching their first course Chapters are written both by established experts in the field and by new professors within their first couple of years of teaching

Second Revision of the Bibliography of Educational Measurements Indiana University. School of Education. Division of Research and Field Services, Henry Lester Smith, Wendell William Wright, 1927 Viewpoints ,1928 **Report of the Proceedings and Addresses of the ...** National Catholic Educational Association,1927 NCEA Bulletin National Catholic Educational Association,1927 **Bulletin** National Catholic Educational Association,1928 **Contributions to Education** New York Society for the Experimental Study of Education,1928 *Diversity, Equity, and Inclusion for Mathematics and Science Education: Cases and Perspectives* Lin, Cheng-Yao, Sun, Li, 2025-06-04 Diversity equity and inclusion DEI are critical pillars for transforming mathematics and science education As classrooms diversify the need to address systemic barriers and create inclusive learning environments becomes more urgent Cases on DEI in STEM education highlight the real world challenges and strategies educators face in promoting equitable access to learning opportunities dismantling biases and empowering students from historically marginalized communities Further exploration may reveal powerful teaching tools and catalyze reflective practice and institutional change encouraging educators to critically examine their roles in shaping a more inclusive future in math and science Cases on Diversity Equity and Inclusion for Mathematics and Science Education explores key issues and concepts related to diversity equity and inclusion in mathematics and science classrooms It offers solutions and successful strategies for teaching and learning in mathematics and science This book covers topics such as inclusive classrooms K 12 education pre service teaching and is a useful resource for educators sociologists academicians researchers and scientists How to Measure Guy Mitchell Wilson, Kremer Jacob Hoke, 1928

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