

For each diagram below, decide whether entropy will increase or decrease and explain your answer.

Entropy



Entropy will increase
(increase/decrease)

Entropy increases when molecules become a gas because they are more disordered and energy is spread over a larger area.



Entropy will decrease
(increase/decrease)

Entropy decreases when more order is established in a system and solids are more highly ordered than liquids.



Entropy will increase
(increase/decrease)

When temperature increases, gas molecules move even faster, colliding faster with each other, and distributing energy faster.



Entropy will increase
(increase/decrease)

When solids dissolve, the molecules spread out over a larger area and energy is distributed over a wider area, causing entropy to increase.



Entropy will increase
(increase/decrease)

Entropy tends to increase when

a) a solid is no longer a solid at the end of the reaction

In this case, the reactant Zn is a solid, but there is no



Entropy will decrease
(increase/decrease)

In this reaction, a very highly ordered solid is

produced, which has lower entropy. Also, 6 moles of liquid and 6 moles of gas combine to form a mole of

Teacher Answer Key Included!

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Bernd Finkbeiner, Leonardo Mariani



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 - Advantages of eBooks Over Traditional Books
2. Identifying Entropy Question And Answers Download
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Entropy Question And Answers Download
 - User-Friendly Interface
4. Exploring eBook Recommendations from Entropy Question And Answers Download
 - Personalized Recommendations
 - Entropy Question And Answers Download User Reviews and Ratings
 - Entropy Question And Answers Download and Bestseller Lists
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 - Interactive Elements Entropy Question And Answers Download
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Entropy Question And Answers Download
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Entropy Question And Answers Download
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Entropy Question And Answers Download
 - Setting Reading Goals Entropy Question And Answers Download
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Entropy Question And Answers Download
 - Fact-Checking eBook Content of Entropy Question And Answers Download
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
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 - Integration of Multimedia Elements

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der häftlinge und brachte sie ins hotel pragser wildsee wo sie am 4 mai 1945 von amerikanischen truppen übernommen

wurden grundlage für das doku drama wir geiseln der ss rezensionen