

Experiment 8 Report Sheet

Limiting Reactant

Date _____

A. Precipitation of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ from the Salt Mixture

Unknown number B

1. Mass of beaker (g)
2. Mass of beaker and salt mixture (g)
3. Mass of salt mixture (g)
4. Mass of filter paper (g)
5. Mass of filter paper and product after air-dried or oven-dried (g)
6. Mass of dried product (g) 1.191
7. Formula of dried product

B. Determination of Limiting Reactant

1. Limiting reactant in salt mixture (write complete formula)
2. Excess reactant in salt mixture (write complete formula)

Data Analysis

1. Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (or CaC_2O_4) precipitated (mol)
2. Moles of limiting reactant in salt mixture (mol)
• formula of limiting hydrate $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$
3. Mass of limiting reactant in salt mixture (g)
4. Mass of excess reactant in salt mixture (g)
• formula of excess hydrate $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$
5. Percent limiting reactant in salt mixture (%)
6. Percent excess reactant in salt mixture (%)
7. Mass of excess reactant that reacted (g)
8. Mass of excess reactant, unreacted (g)

*Show calculations for Trial 1 on next page.

Trial 1 250 mL

Trial 2

96.553g

47.956g

97.559g

49.083g

1.001g

1.067g

0.785g

0.785g

1.976g

1.774g

1.191g

0.989g

$\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$

$\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$

$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$

$\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$

0.00815 mol

0.00676 mol

2. Moles of limiting reactant in salt mixture (mol)

• formula of limiting hydrate $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$

0.00815 mol

0.00676 mol

3. Mass of limiting reactant in salt mixture (g)

1.198g CaCl_2

0.994g CaCl_2

4. Mass of excess reactant in salt mixture (g)

• formula of excess hydrate $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$

0.192g

0.073g

5. Percent limiting reactant in salt mixture (%)

83.97%

93.16%

6. Percent excess reactant in salt mixture (%)

19.32%

6.84%

7. Mass of excess reactant that reacted (g)

1.4922g

1.2343g

8. Mass of excess reactant, unreacted (g)

1.3030g

1.1606g

$$\textcircled{D} \quad \frac{1.191g \text{ CaC}_2\text{O}_4}{146.12g} \times 1 \text{ mol}$$

$$\frac{0.989g}{146.12g} \times 1 \text{ mol}$$

Experiment 8 Limiting Reactant Answers

Anthony S. Fauci



Experiment 8 Limiting Reactant Answers:

Adopting the Track of Phrase: An Emotional Symphony within **Experiment 8 Limiting Reactant Answers**

In some sort of taken by monitors and the ceaseless chatter of fast transmission, the melodic elegance and psychological symphony developed by the published term often diminish into the back ground, eclipsed by the constant noise and interruptions that permeate our lives. However, situated within the pages of **Experiment 8 Limiting Reactant Answers** a marvelous literary value filled with raw emotions, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, this interesting masterpiece conducts readers on an emotional trip, well unraveling the concealed tunes and profound influence resonating within each carefully constructed phrase. Within the depths of this emotional review, we can investigate the book is main harmonies, analyze its enthralling publishing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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