

Submitted by:

Skyline High School in

Sammamish, Washington

Tv Swiftney

Thomses to:

Ward's Science

# Lab: Dynamic Equilibrium Simulation

## FOR THE TEACHER

#### Summary.

In this lab, students will explore equilibrium using paper clips to mimic a chemical reaction.

#### Resource Type Lab

Grade Level High School

#### Objectives

By the end of this lesson, students should be able to

Better understand what it means for a system to reach equilibrium.

#### Chemistry Topics

This lesson supports students' understanding of

- Equilibrium
- Le Châtelier's Principle

#### Time

Teacher Preparation: 30 minutes Lesson: 1 hour

#### Materials

For each group:

- -60 paper clips
- timer

## Safety

No notable safety precautions need to be observed for this lab.

## Teacher Notes

- Students should work in groups of four: Two people will put paper clips together, one
  person will take paper clips apart, and one person will be the timer and record data.
- Students are required to generate their own data table for the three parts.

## FOR THE STUDENT

## Student Activity Sheet: Dynamic Equilibrium Simulation Lab

## Lesson

## Background

You will be modeling the reaction: P + P = P₂
by putting paper clips together (forward reaction) and taking them apart (reverse reaction). This simulation allows you to "stop" the reaction after a certain amount of time so that you can monitor the "concentrations" of the reactants and products.

# **Dynamic Equilibrium Lab Manual**

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