

Name: \_\_\_\_\_  
Period: \_\_\_\_\_

## Fall Semester Review – IPC

IPC Chemistry Review

Put in the table at the right from your periodic table.

Element	Chemical Symbol	Atomic #	Valence Electrons	Oxidation #	# of Protons	Atomic Mass	Metal or Nonmetal	Group or Luster Electrons	# of Full Electron Levels
Oxygen									
Argon									
Calcium									
Nitrogen									
Potassium									
Sulfur									
Helium									

Find the molecular mass of the following compounds

Calcium oxide (CaO)

Water

Classify (ionic, covalent or polyatomic) and name the following compounds.

1. CO<sub>2</sub> \_\_\_\_\_
2. Na<sub>2</sub>O \_\_\_\_\_
3. BaCrO<sub>4</sub> \_\_\_\_\_

Draw the Lewis Dot Diagrams for

1. Carbon
2. Sodium

Make balanced ionic compounds from:

1. Li + O \_\_\_\_\_
2. Na + Ar \_\_\_\_\_
3. Be + NO<sub>3</sub><sup>-</sup> \_\_\_\_\_
4. Ca + O \_\_\_\_\_

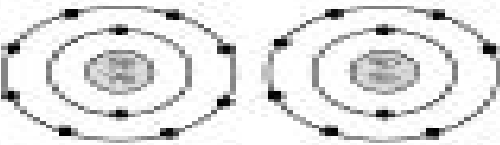
With dot diagrams draw the covalent bond of O<sub>2</sub>

With dot diagrams the covalent bond of CH<sub>4</sub>

1. Valence electrons are the outermost electrons of an atom that are involved in chemical bonding? True/False

2. Lithium has \_\_\_\_\_ valence electrons, will lose or gain \_\_\_\_\_ electrons and become \_\_\_\_\_ (positive or negative). This is why lithium's oxidation number is \_\_\_\_\_.

3. A metal and non-metal will form a \_\_\_\_\_ compound. Two non-metals form a \_\_\_\_\_ compound.



Atom A
Atom B

Which one is the ion?

Give the ion notation for the ion.

**Matching:**

- A. Ion
- B. Element
- C. Isotope
- D. Oil

If you change the number of protons, you change the \_\_\_\_\_.

If you change the number of neutrons, you change the \_\_\_\_\_.

If you change the number of electrons, you change the \_\_\_\_\_.

Every 3,000 miles you should change a car's \_\_\_\_\_.

**Matching:**

- A. Law of Conservation of Mass
- B. Meter
- C. Centimeter
- D. Liter
- E. Gram

Unit of length: equal to 3.2 ft.  
Mass is neither destroyed or created in chemical reactions.  
Unit of mass: about 1 dollar bill.  
Unit of mass: width of pinky finger.  
Unit of volume: just bigger than a quart.

**Matching:**

- A. Solid
- B. Liquid
- C. Gas
- D. Oil

Molecules that are tightly packed and retain their shape and size.  
Molecules that bounce off of each other, can be compressed, and take the shape of their container.  
Molecules that can move (slide) over each other and have a definite size (volume), but not shape and can not be compressed.

**Matching:**

- A. Mixture
- B. Compound
- C. Element
- D. Matter
- E. Heterogeneous
- F. Homogeneous

A substance made up of two or more atoms that must be separated by chemical means.  
Something made up of 2 or more different atoms that can be separated by physical means.  
A substance that is pure and made up of only one type of atom.  
Anything that has mass and takes up space.  
A mixture that is the same throughout.  
A mixture that is different throughout.

# Fall Semester Review Ipc Answers 2014

**Examreview**



## **Fall Semester Review Ipc Answers 2014:**

**Volcanic Ash** Shona Mackie,Katharine Cashman,Hugo Ricketts,Alison Rust,Matt Watson,2016-05-24 Volcanic Ash Hazard Observation presents an introduction followed by four sections each on a separate topic and each containing chapters from an internationally renowned pool of authors The introduction provides a volcanological context for ash generation that sets the stage for the development and interpretation of techniques presented in subsequent sections The book begins with an examination of the methods to characterize ash deposits on the ground as ash deposits on the ground have generally experienced some atmospheric transport This section will also cover basic information on ash morphology density and refractive index all parameters required to understand and analyze assumptions made for both in situ measurements and remote sensing ash inversion techniques Sections two three and four focus on methods for observing volcanic ash in the atmosphere using ground based airborne and spaceborne instruments respectively Throughout the book the editors showcase not only the interdisciplinary nature of the volcanic ash problem but also the challenges and rewards of interdisciplinary endeavors Additionally by bringing together a broad perspective on volcanic ash studies the book not only ties together ground air academic and applied approaches to the volcanic ash problem but also engages with other scientific communities interested in particulate transport Includes recent case studies highlighting the impact of volcanic ash making methods used for observation more accessible to the reader Contains advances in volcanic ash observation that can be used in other remote sensing applications Presents a cross disciplinary approach that includes not only methods of tracking and measuring ash in the atmosphere but also of the fundamental science that supports methodological application and interpretation Edited by an internationally recognized team with a range of expertise within the field of volcanic ash

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